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0620/06/M/J/03 4 A mixture of two solid compounds D and E was analysed. Solid D was a zinc salt which is soluble in water. Solid E was an insoluble metal carbonate. The tests on the mixture and some of the observations are in the following table. Complete the observations in the table. For Examiner's Use tests observations (a) About half of the mixture of D and E

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Can somebody please help me with this Q When 0.47 g of E was completely burnt in air the heat produced raised the temperature of 200 g of water by 27.5°C assume no heat losses occurred during this experiment bi) use the relevant data from the data booklet to calculate the amount of heat released...

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0620 Chemistry June 2015 Principal Examiner Report for Teachers © 2015 CHEMISTRY Paper 0620/21 Core Theory Key Messages It is important that candidates read questions carefully in order to understand what is exactly being asked. Many candidates would benefit from more practice in answering questions about the properties of Group

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